
1. Identification of the substance/preparation

Road Salt

CAS nr: 7647-14-5

EG-nr: 231-598-3

Other name: Sodium chloride (NaCl)

Distributor: Salinity UK Ltd
PO box 227 Didcot
Oxon OX1 11 AB

Tele: +44 1235 832 086
Fax: +44 787 635 0917

Emergency Advice: Poisons information center 112

2. Composition/Information on ingredients

<i>Name of component</i>	<i>CAS-nr</i>	<i>concentration (%)</i>		
Sodium Chloride	7647-14-5	98 till 100%	-	-

3. Possible Hazards

Not classified as a dangerous product according to the EC-guideline in the latest version.

4. First aid measures

General Information – Consult physician with continuous complaints.

In case of inhalation – In case of inhalation ensure fresh air.

In case of skin contact – In case of skin contact rinse off with water.

In case of eye contact – In case of eye contact rinse thoroughly with water.

After swallowing – Rinse mouth with water and drink plenty of water.

5. Fire-fighting measures

The product is not combustible.

In the event of fire: Hydrogen Chloride (HCL) can be set free.

In the event of fire do not breath fumes.

Collect contaminated fire fighting water separately, may not be discharged into the drains.

6. Measures during unintentional release

Avoid dust formation.

Pick up mechanically.

Do not discharge into the drains or bodies of water.

7. Handling and Storage

Advice on safe handling

No special measures necessary if used correctly.

Storage

Sodium Chloride dissolves in water, thus it should be stored in a dry place.
Sodium Chloride and water is corrosive on most metals.
Sodium Chloride is stable during storage.

8. Exposure controls / personal protection

Avoid dust formation.

In case of dust formation wear goggles, micro mask and protective gloves.

9. Physical and chemical properties

<i>Description</i>	Crystalline, colorless, odorless
<i>Chemical name</i>	NaCl
<i>Molvikt</i>	58,4 g/mol
<i>Densitet</i>	2 160 kg/m ³ , bulk density 1 100 till 1 400 kg/m ³
<i>Melting point</i>	801 °C
<i>Boiling point</i>	1413 °C
<i>Vapor pressure</i>	1 mm Hg at 865 °C
<i>Solubility in water</i>	317 gram per liter at 20 °C
<i>pH</i>	neutral

10. Stability and reactivity

Stable under normal conditions.

Reactions with strong acids- Hydrogen chloride (HCL)

Reactions with oxidizing agents – chlorine gas (Cl₂)

Corrosive to metals.

11. Toxicological information

Acute toxicity – oral

LD50 oral rat over 3 000 mg/kg body mass

Acute toxicity - dermal

LD50 dermal rabbit over 10 000 mg/kg body mass

12. Ecological information

Acute toxicity

LC50 fish 96 h	> 7 500 mg/l water
LC50 Daphnia 48 h	> 1 000 mg/l water
EC50 algae (nitszcheria lineris) 120 h	2 430 mg/l

Chronic toxicity

Fish (Salmo Gairdneri) NOEC mortality, 192 h	800 mg/l
Daphnia, LOEC, reproduction, 21 24-h periods	1 714 mg/l

Mobility

In air	As an aerosol of solid particles
In water	Sodium Chloride dissolves easily in water

Degradation

Sodium Chloride is split in Na⁺ and Cl⁻ in water. .

Remark: Sodium Chloride is a naturally occurring mineral in soil and water.

13. Disposal considerations

In accordance with local authority regulations

EWC-kod 06 03 14 Solid salts and solutions except those mentioned in 06 03 11 and 06 03 13.

Uncontaminated packaging may be taken for recycling. Contaminated packaging should be emptied as far as possible and discharged like product.

14. Transportation information

No hazardous material as defined by the regulations.

15. Regulatory information

The product is not classified as dangerous according to the European Chemical legislation, thus there is no demand for a product information sheet.

16. Regulatory information

The product is not for human consumption.

The information given is based on our present knowledge and is not meant to guarantee product properties. Recipients of our product must take responsibility for observing existing laws and regulations.

Refer to product information.